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1- every sample has exact same characteristic physical and chemical properties used to identify substance 2- every sample of a pure substance has exact same composition with the same makeup (i.e.: pure water is always 11.2% H and 88.8% oxygen by mass

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This topic covers 1.2 Bonding and Structure and 1.3.1 Periodicity (melting and boiling trends only) Bonding, Structure and Periodicity hw (mark scheme) Bonding, Structure and Periodicity test (mark scheme) More questions on Bonding, Structure and Periodicity (mark scheme) 1.2.2 Exercise 1 - ionic bonding and ionic compounds

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Textbook Answers | GradeSaver

Chemistry Chapter 1 Test Review Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank. \_\_\_\_ 1. Inorganic chemistry is the study of a. non-carbon related compounds. b. the chemistry of living things. c. mathematical modeling. d.

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2 ANSWERS TO EXAM-STYLE QUESTIONS 1 CHEMISTRY FOR THE IB DIPLOMA © CAMBRIDGE UNIVERSITY PRESS 2011 molar mass of I 2 = 253.80 g mol<sup>-1</sup> mass of I 2 = 1.25 × 10<sup>13</sup> × 253.80, i.e. 0.317 g 16 a molar mass of PbI 2 = 460.99 0.1270 moles of PbI 2 = 460.99 = 2.755 × 10<sup>14</sup> mol b Pb(NO 3) 2(aq) + MI 2(aq) ⇌ PbI 2(s) + M(NO 3) 2(aq)

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