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the mass of the  
solute dissolved,  
and  $V$  is the total  
volume of the  
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palladium may be  
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exposing Pd metal  
to H<sub>2</sub> gas. The

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concentration of hydrogen in the palladium depends on the pressure of  $H_2$  gas applied, but in a more complex fashion than can be described by Henry's law. Under certain conditions, 0.94 g of hydrogen gas is dissolved in 215 g of palladium

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million ( $10^6$ ) grams  
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And Obj Answers

- A. Buffer solution
- B. Neutral solution
- C. Saturated solution
- D. Supersaturated solution.

10.

Protein is a polymer formed from the linkage of

- A. Amino acid molecules.
- B. Fatty acid molecules.
- C. Glucose units.
- D.

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SECTION A. 1.

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remaining when  
each of the  
following is heated.

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