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MULTIPLE CHOICE

On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question.

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ANSWER Answer the following questions in the space provided. 1. a Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not

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involve chance
discoveries. (c) is driven
by curiosity. (d) is done
for the sake of learning
something new.

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SECTION 3 SHORT

ANSWER Answer the
following questions in
the space provided.

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thing. Terms in this set
(25) Chemical formula.
indicates the relative
number of atoms of
each kind in a chemical
compound. Reveals the
number of atoms of

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each element contained
in a single molecule ...

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CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the

following questions in

the space provided. 1. c

In the modern periodic

table, elements are

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Key

ordered (a) according to decreasing atomic mass.

(b) according to Mendeleev's original

design. (c) according to increasing atomic

number. (d) based on when they were

discovered. 2. d

Mendeleev noticed that certain similarities in the

...

5 The Periodic Law

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CHAPTER 13

REVIEW Ions in Aqueous Solutions and Colligative Properties

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 1 on page 437 of the text to predict the solubility of the following compounds in water: soluble a.

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magnesium nitrate

insoluble b. barium

sulfate insoluble c.

calcium ...

Key

13 Ions in Aqueous

Solutions and

Colligative Properties

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CHAPTER 6 REVIEW
Chemical Bonding
SECTION 1 SHORT
ANSWER Answer the
following questions in

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the space provided. 1. a

A chemical bond
Chapter 1
Review Answer
between atoms results
Key
from the attraction

between the valence
electrons and of

different atoms. (a)

nuclei (c) isotopes (b)

inner electrons (d)

Lewis structures 2. b A

covalent bond consists

of (a) a shared electron.

6 Chemical Bonding

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the

following questions in
the space provided. 1.

Given the following
equation: $C_3H_4(g) +$
 $xO_2(g) \rightarrow 3CO_2(g) +$
 $2H_2O(g)$ 4 a. What is

the value of the
coefficient x in this
equation? 40.07 g/mol
b. What is the molar

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mass of C₃H₄? 2 mol
O 2:1 mol H₂O c. What
is the mole ratio ...

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Chapter 5 Review. by

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